	Table of Contents Introduction	2
	Objectives	2
	Pre-required skills	2
	Materials	2
	Example 1. Analysing a solvate structure	3
Solvate Analyser	Solvent selection and space calculation	
(MAT-003)	Investigating H-bonding	6
	Generating summary reports	6
2020.3 CSD Release	Example 2. Exploring disordered solvate structures	7
	Conclusions	9
	Feedback	9



# Introduction

This tutorial will introduce you to the *Solvate Analyser* in *CSD-Materials*.

The Solvate Analyser tool in Mercury provides the user with the ability to quickly investigate complex solvated structures.

Before beginning this workshop, ensure that you have a registered copy of CSD-Materials or above installed on your computer. Please contact your site administrator or workshop host for further information.

### Objectives

- Explore visualisation options
- Solvent selection and space calculation
- Investigate hydrogen bonding motifs in solvate structures
- Analyse disordered solvents

This workshop will take approximately **1** hour to be completed.

# Pre-required skills

Familiarity with the Mercury interface is important; you can access the Visualization in Mercury self-guided workshop <u>here</u>. You can find other CSD-Materials self-guided workshops <u>here</u>.

# Materials

There are no additional materials required for this workshop.





# Example 1. Analysing a solvate structure

The physicochemical properties of crystalline materials, such as pharmaceutical compounds and excipients, can be considerably influenced by the presence of solvent molecules within the crystal structure. Solvents are frequently found in the crystal structures of pharmaceutical compounds, either interacting directly with the host compound or simply filling gaps in the structure. Understanding the behaviour of solvents in these structures is of great importance in pharmaceutical industry since this can lead to further understanding and improvement of the crystallization process.

This example shows how to assess solvate structures using the Solvate Analyser within CSD-Materials. The case studies illustrate how to quickly investigate complex solvated structures, including structures with more than one solvent, a mixture of solvents, co-formers and counter-ions or even disordered solvate structures.



2



6-(4-Ammoniophenyl)-4,5-dihydro-5methyl-3(2H)-pyridazinone (-)-tartrate dichloromethane methanol solvate salt CSD refcode **HAVKIY** 

# Solvent selection and space calculation

In this section, you will learn how to generate and manipulate solvent space in a salt containing a mixture of solvents.

- 1. Open Mercury by clicking the desktop icon, or launching from the Start menu (Windows), Launchpad (macOS) or command line (Linux).
- 2. In the **Structure Navigator** window, type the refcode *HAVKIY*, to bring up the structure of 6-(4-Ammoniophenyl)-4,5-dihydro-5-methyl-3(2H)-pyridazinone (-)-tartrate dichloromethane methanol solvate.
- 3. The structure will be displayed in the 3D visualiser.
- 4. From the top-level menu select **CSD-Materials > Solvate Analyser**....



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	CSD-Mater	rials Wizard		
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	Polymorph	Assessment	•	
	Co-Crystal	Design	•	
	Full Interac	tion Maps		
	Hydrate Ar	nalyser		
	Solvate An	alyser		
	Conforme	r Generation	13	
	Launch DA	SH		

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- 5. The Solvate Analyser contains 3 tabs with different functionality options. In the Solvent Selection and Space Calculation tab you can individually select different solvent molecules within the structure and analyse them independently. Hold shift and click on the dichloromethane in the 3D visualiser. Alternatively, you can select the molecule by using the Lasso selection mode from the top-level toolbar. With the dichloromethane molecule selected, click on Add Solvent From Selected button to register that group of atoms as a specified solvent.
- 6. Information is now displayed in the table for dichloromethane. Repeat the same steps for methanol: select the molecule and click **Add Solvent From Selected**.
- 7. Methanol information is now added in second column of the table. Doubleclick on each cell in the **Solvent** row, which is interactive, and add the names for each solvent.

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#### 👽 s olvate Analyse Solvent Selection and Space Calculation Solvent H-Bonding Structure Summary Hide all molecules Packing... Add Solvent From Selected Remove Solvent Calculate Space 2 Settings Solvent dichloromethan methanol Probe Radius: 1.2 Approx. Grid Spacing: Formula C1 H2 CI2 C1 H4 O1 0.3 Calculate using the Contact Surface Volume (%) Volume (Å<sup>3</sup>) Results 0.0 % of unit cell volume $\checkmark$ $\checkmark$ Show Space Volume 0.00 Å<sup>3</sup> $\checkmark$ $\checkmark$ Show Solvent Select Solvent Defaults Inside Colour 0.45 Inside Opacity 0.45 Outside Colou Outside Opacity 0.45 0.45 Close

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- 8. You can make the solvents easier to see by clicking on the **Select Solvent** boxes and change the *Style* to **Ball and Stick** from the top-level Mercury toolbar. Click **Calculate Space** to generate the solvent space using the default parameters. This will automatically create a 1x1x1 packing and display the space occupied by solvents or structural voids.
- 9. You can change the inside and outside colour of the solvent space. For instance, click on the **Inside Colour** of dichloromethane and select the deep yellow, then click on the **Outside Colour** and select light yellow.
- 10. To further explore the location of the solvent space click the **Packing...** button. Enter 0.0 to 2.0 for *a* axis and -0.5 to 2.0 for *b* and *c* axes to obtain a more complete picture of the voids. Click **OK** and then **Calculate Space**. Click *b* axis from the top-level toolbar to view along it. Note that the dichloromethane solvent forms channels along the *a* axis, while the methanol solvent is located in discrete pockets. Select the view along *a* axis. A double column of dichloromethane molecules are filling the channel space and one methanol molecule is occupying the discrete pockets.

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Solvent Selection and Space Calculation Solvent H-Bonding Structure Summary





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#### Investigating H-bonding

In this section you will learn how to quickly analyse solvent H-bonding motifs. To go back to the initial view click **Reset** in the *Display Options* area of the Mercury interface

- 11. Check on the **Solvent H-Bonding** tab. This shows an automated analysis of the H-bonding environments of the solvate molecule(s). This is a quick way to investigate the H-bonding in complex solvate structure. The methanol molecule is involved in two H-bond interactions with the tartrate. There is one donation and one acceptance each from O8 of the methanol molecule.
- 12. To identify the atoms involved in H-bond interactions toggle on the **H-Bond** check box in the *Display Options* dialogue box.
- 13. Display the atoms labels and expand the contacts by clicking on the hanging atoms (red dotted line).

#### Generating summary reports

14. To generate solvent analysis reports, select **Structure Summary** tab in the *Solvent Analyser* dialogue box. You can customise the information that is displayed in this tab and output in your report by using the **Configure...** button. Once you are happy click on **Save As...** button to save a report in Word (DOCX) or HTML format.



	~	nd Space Calculation	Solvent H-Bonding	_						
Nu	umber	Donor mol.	Acceptor mol.	Donor atom	H atom	Acceptor atom	Length	Length-VdW	Symm. op. D	Symm. op. A
	1	methanol	non-solvent	<b>0</b> 8	III H24	O2	1.827	-0.893	x, 1+y, z	1+x,1+y,z
Г 2	2	non-solvent	methanol	<b>0</b> 4	III H16	O8	1.746	-0.974	1+x,1+y,z	1+x, 1+y, z

olvent Selection and	Space Calculation	Solver	nt H-Bonding	Structure S	ummary				
Identifier	HAVKIY								
Compound Name	(R)-(-)-6-(4-Ammo	oniopheny	l)-4,5-dihydro-9	5-methyl-3(2H	l)-pyridazinone	(-)-tartrate	dichlorometh	ane methanol solvate	:
Chemical Diagram				O1= HN HO8 C	11 T 07	$\sim$			
Solvent H-bonds	Number Donor 1 metha 2 non-so	anol n	ceptor mol. I on-solvent methanol	<b>Donor atom</b> O8 O4	Hatom Acc H24 H16	optor ato O2 O8	<b>m Length I</b> 1.827 1.746	L <b>ength-VdW</b> -0.893 -0.974	
Solvent in Donus			Volume(ų)	Volume (%	) Probe Radi	us (Å)			
Solvent volume	dichloromethan C		482.67 223.11	21.0 9.7	1.2				

# Example 2. Exploring disordered solvate structures

One advantage of using the Solvate Analyser tool is the ability to quickly generate solvent space for complex, disordered solvents.

In this example you will learn how to display solvent space for disordered structures.





2'-(indol-3-yl)-2-oxospiro(indoline-3,4'-pyran) derivative ethanol methanol solvate **PACMIR** 

- 1. Open Mercury by clicking the desktop icon, or launching from the Start menu (Windows), Launchpad (macOS) or command line (Linux).
- 2. In the **Structure Navigator** window, type the refcode *PACMIR*, to bring up the solvated disordered structure.
- 3. From the top-level menu select CSD-Materials>Solvate Analyser....
- 4. In the 3D visualiser select all the methanol atoms then click on Add Solvent From Selected.
- 5. Repeat the same steps for the ethanol molecule.





Structure Na	vigator		6	Y
PACMIR			Find	
Crystal Str	uctures	Spacegroup		^
	PACMIR	P-1		
	PACMIS	P212121		
	PACMOV	C2/c		
	PACMOW	P21/n		
	PACMOX	P21/c		
	DACMOV	D21/n		~
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#### **MAT-003**

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- 6. Type the solvent names into the Solvent row. Toggle on the Select Solvent check boxes and change the *Style* to **Ball and Stick** from the top-level toolbar. Click on Calculate Space to generate the solvent spaces in the crystal lattice.
- 7. To further explore the location of the solvent space click the Packing... button. Enter 0.0 to 2.0 for a and b axes and -0.6 to 2.0 for c axis. Click OK and then **Calculate Space**. Click *b* axis from top-level toolbar to view along it.
- How many molecules of methanol and ethanol occupy the pockets? 8.
- How does the solvent space of the two different molecules intercalate with 9. the parent molecule within the crystal lattice?



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# Conclusions

We have applied the *Solvate Analyser* tool for several structures, including systems with a mixture of solvents and disordered solvent. You should now be able to:

- Select a solvent and calculate the solvent space
- Change the colour of the solvent space
- Investigate hydrogen bonding motifs
- Generate a summary report
- Explore disordered solvate structures

# Feedback

We hope this workshop improved your understanding of the *Solvate Analyser* and you found it useful for your work. As we aim at continuously improving our training materials, we would love to hear your feedback. Click on <u>this link</u> to a survey (link also available from workshops webpage), it will take less than 5 minutes to complete. The feedback is anonymous. You will be asked to insert the workshop code, which for this self-guided workshop is MAT-003. Thank you!